



# APPLICATION OF ION BEAM AND RADIOCHEMICAL TECHNIQUES IN MATERIALS SCIENCE AND ENVIRONMENT

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GREECE**

- Ion-Beam Analysis (IBA) techniques

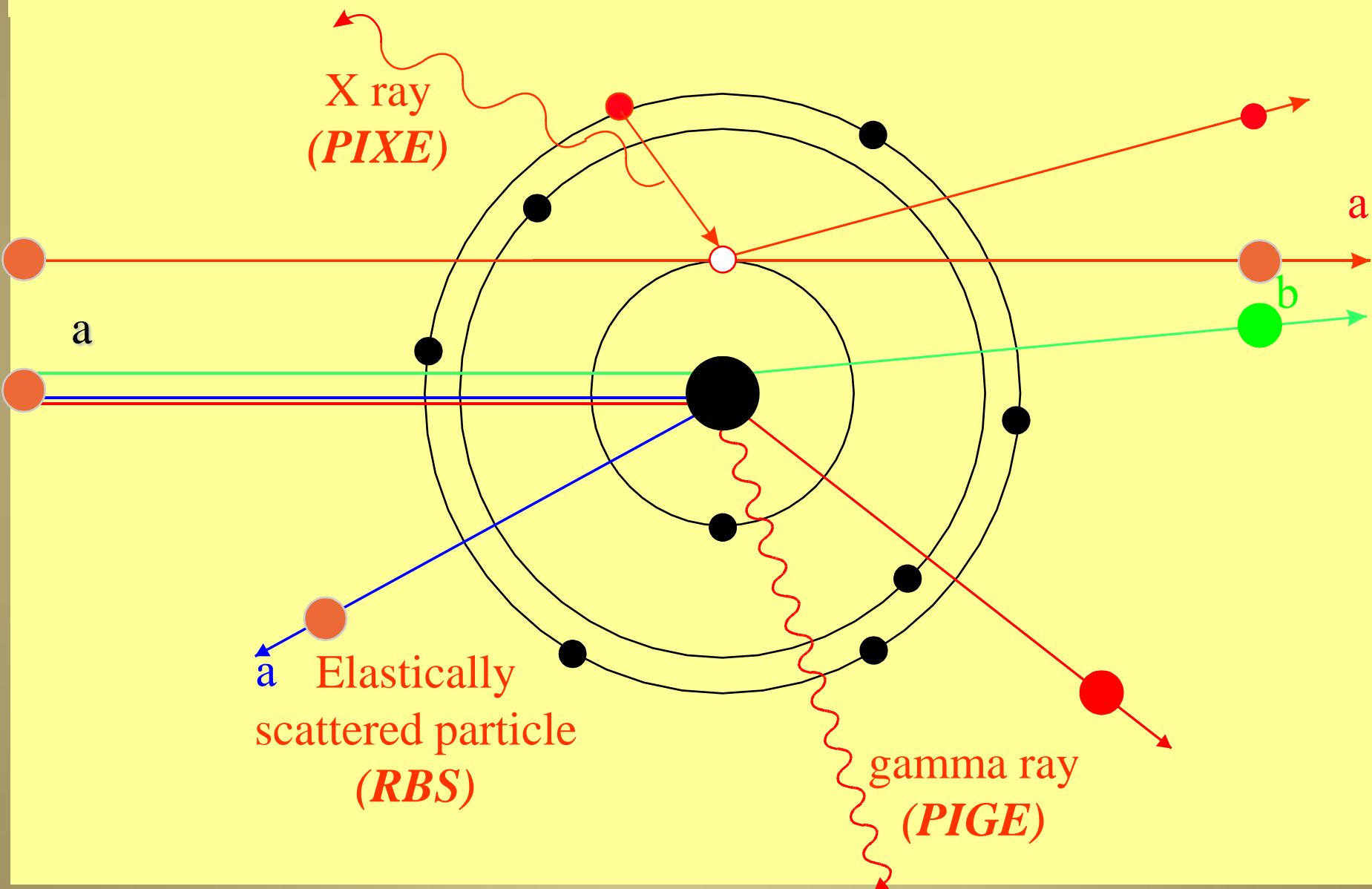
Our aim:

- the characterization of near surface layers of biomaterials in order to investigate their corrosion resistance and biocompatibility
- the characterization and the investigation of the oxidation and corrosion resistance of materials used for industrial applications.

The materials

- Ti-alloys (e.g. Ti-6Al-4V) and Co-based alloys (CoCrMo) used as orthopaedic, dental and cardiac implants
- stainless steels implanted with Al, Zr, Mg, Y for industrial applications
- Cu-alloys in environment and in cultural heritage

# *Ion Beam Analysis*



# A common biomaterial

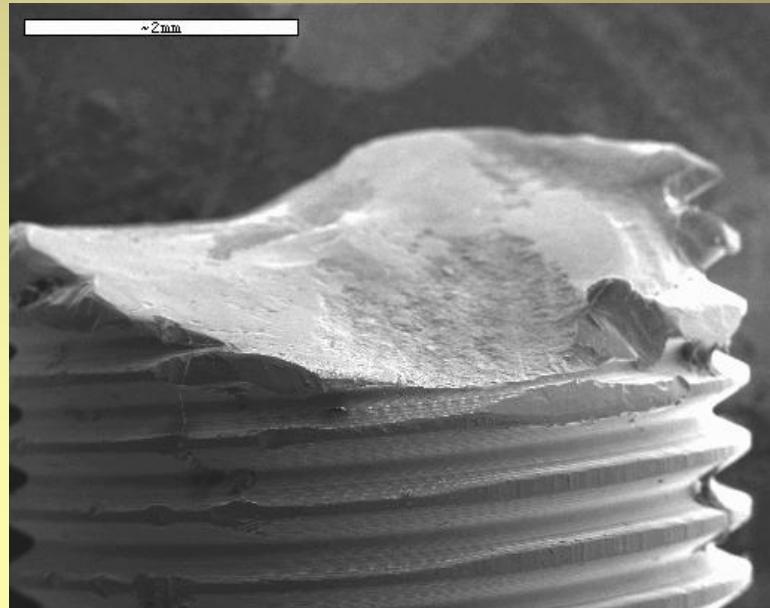
## Ti-6Al-4V

### Advantages

biocompatibility, elasticity,  
corrosion resistance, low  
density

### Disadvantages

friction, hardness,  
wear

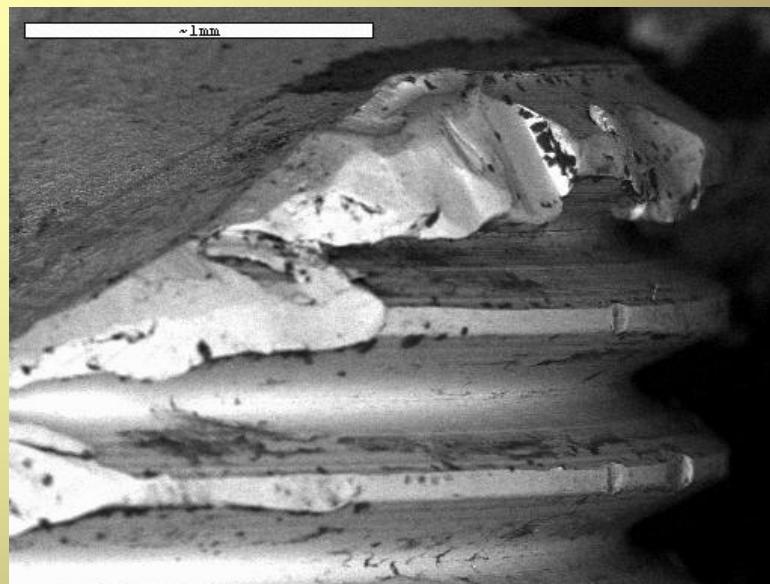


## Surface modification of the material

Ion implantation ,  
plasma nitridation  
or oxidation

### Coatings

SiC (ion beam sputtering)  
+ Ion Beam Mixing,  
CVD, PVD

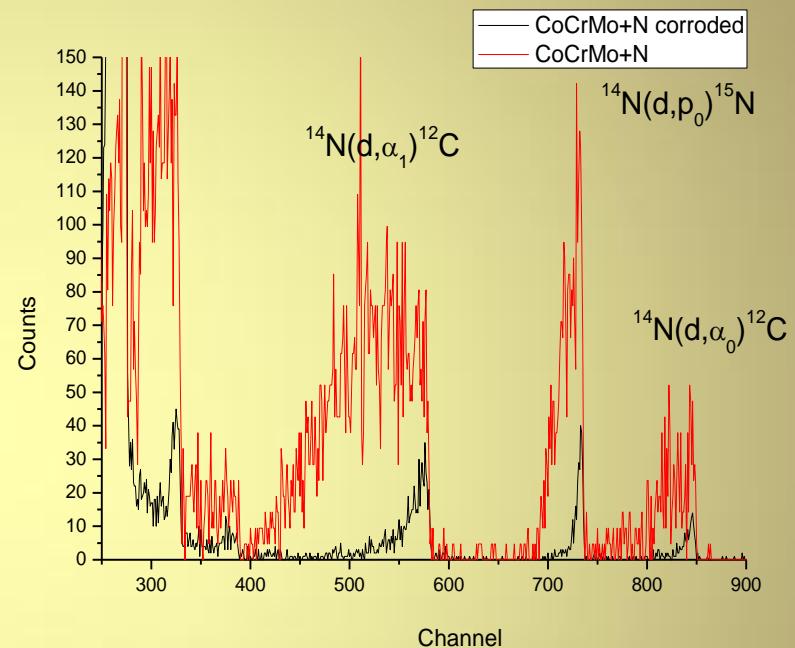
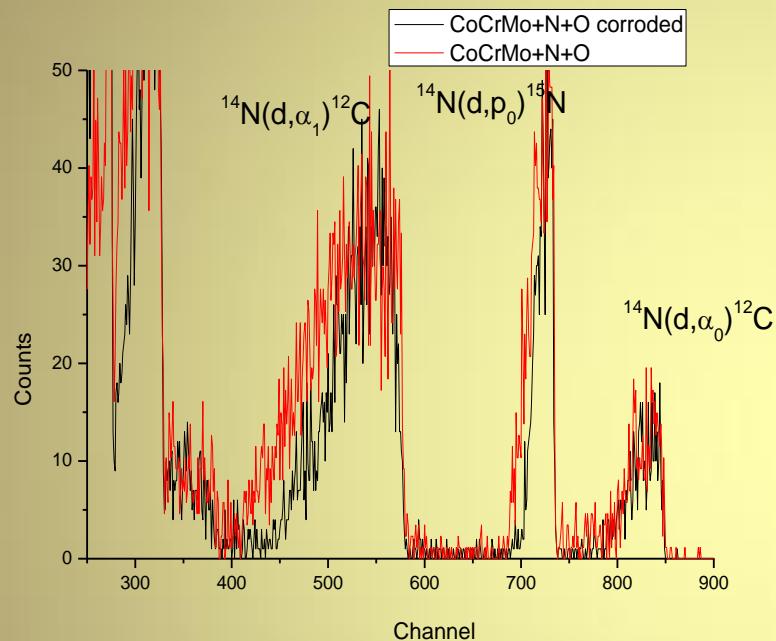


Sample	Treatment	T (°C)	Thickness (μm)	Ti (%)	Ni (%)	N (%)	O (%)	N/Ti
TiNi0*	TiN	300	0.35	48	0	44	8	0.92
TiNi0-N	Nitridation + TiN coating	600-300	0.40	47	0	44	9	0.94
TiNi20	TiN-Ni coating	300	0.43	36	20	36	8	1.00
TiNi20-N	Nitridation + TiN coating	600-300	0.44	36	19	36	9	1.00

\*0, 20 correspond to the Ni-content, N corresponds to the nitrided sample.

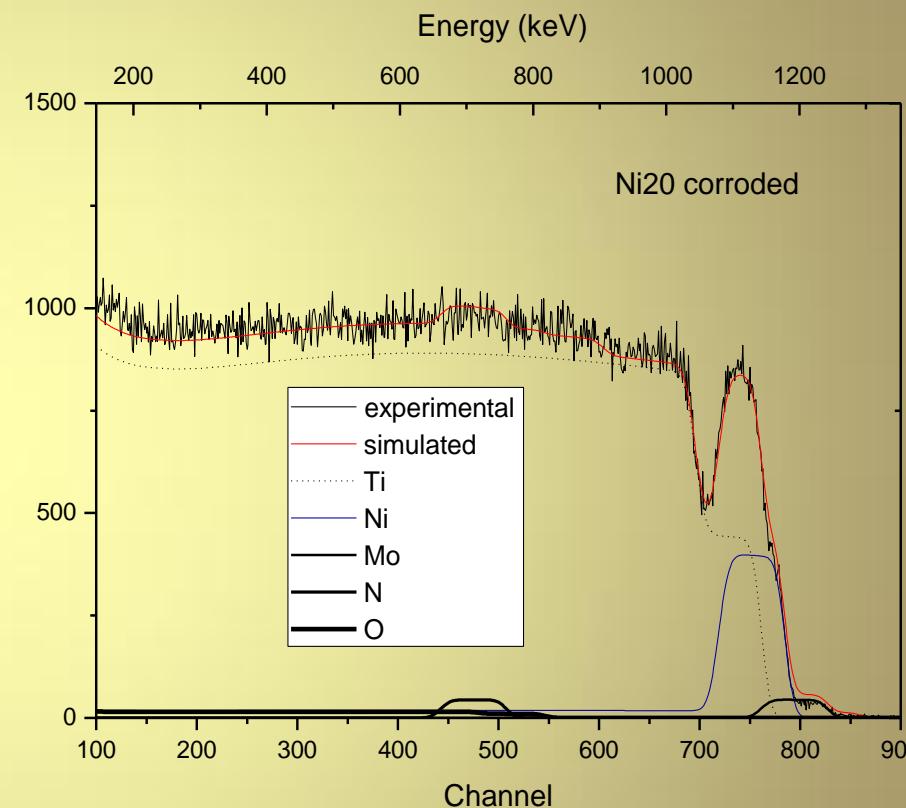
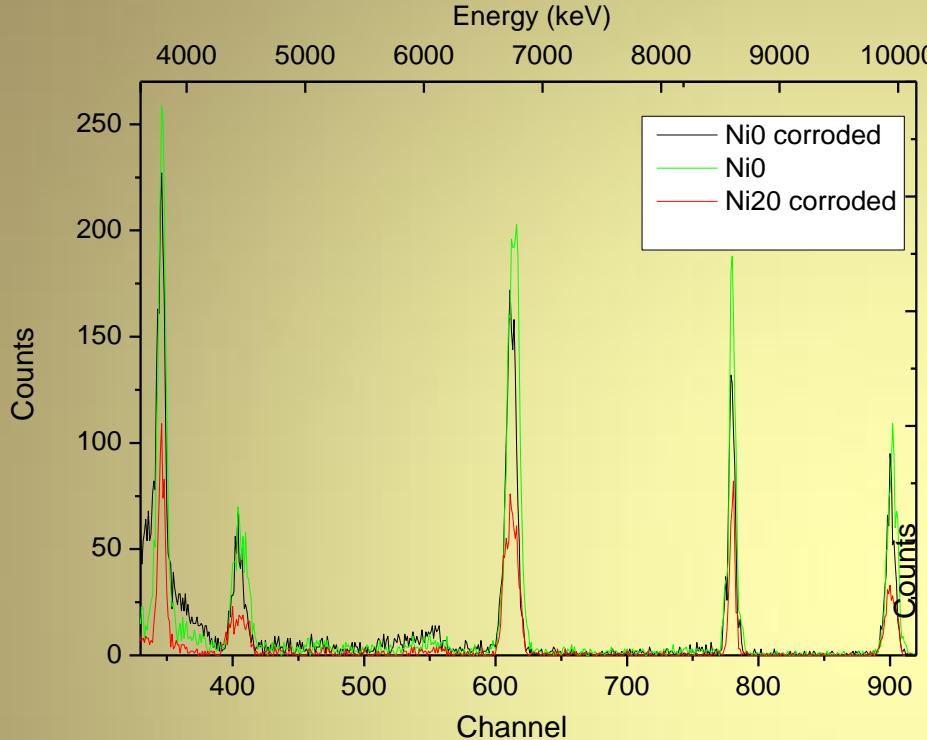
Sample	Treatment	T (°C)	Thickness (μm)	Co(%)	Cr (%)	Mo (%)	O (%)	N (%)
CoCrMo	reference	-	-	60	35	5	-	-
CoCrMo+N	Nitridation	395	4.6	40	20	6	-	34
CoCrMo+O	oxidation	400	0.3	54	20	2	34	-
CoCrMo+ N+O	Nitridation+ oxidation	395 and 400	6.0	40	20	2	15	23

The samples were also investigated prior and after the corrosion tests by d-RBS and NRA ( $E_d = 1.35$  MeV).



The NRA data also proved that the CoCrMo+N+O showed the lowest deterioration and the best corrosion resistance.

## TiN-Ni nano-coating on TAV

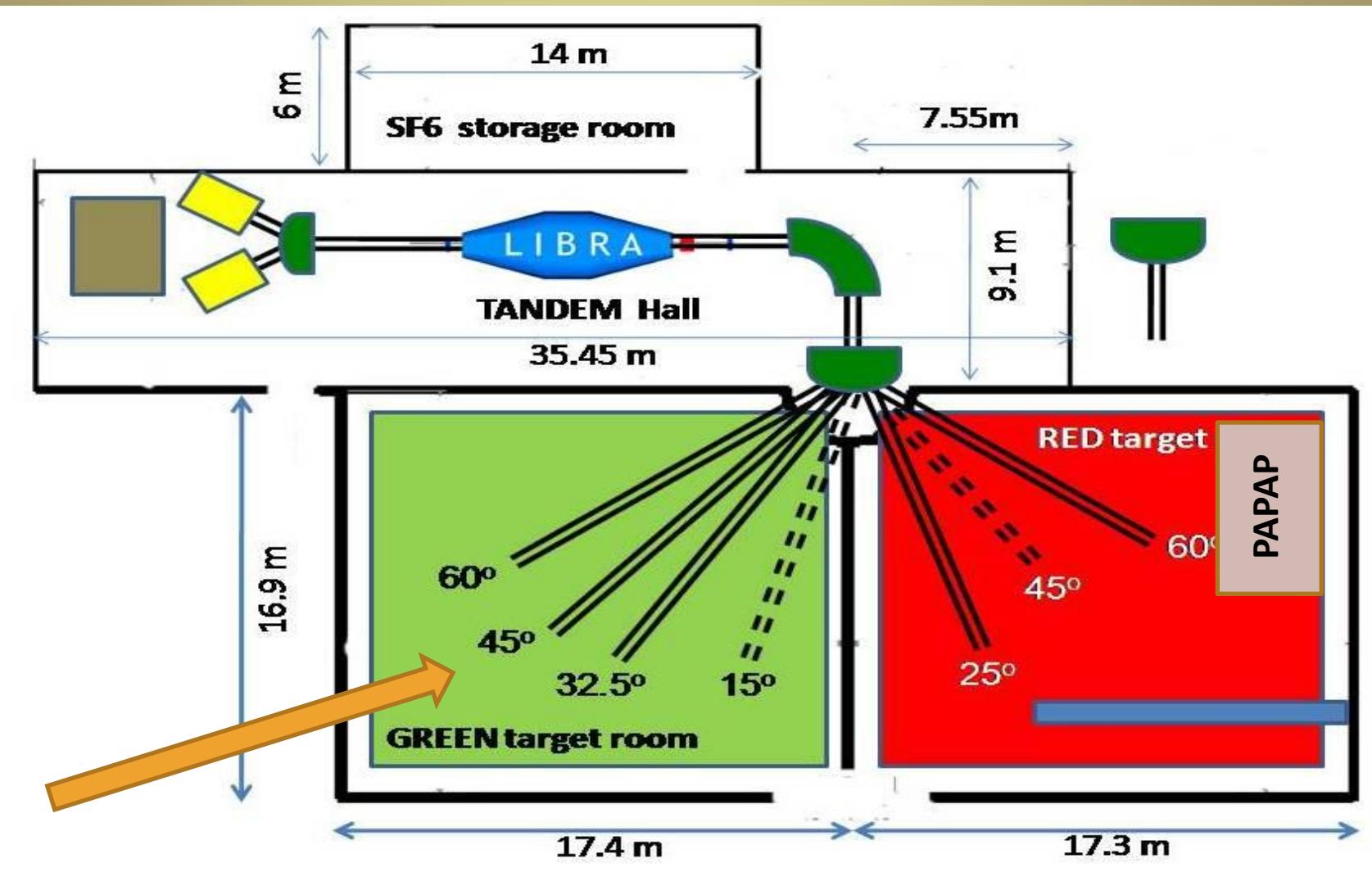


Nitrogen determination by NRA ( $^{14}\text{N}(\text{d},\alpha)$   
and  $^{14}\text{N}(\text{d},\text{p})$  nuclear reactions)

Rutherford Backscattering  
Spectrometry

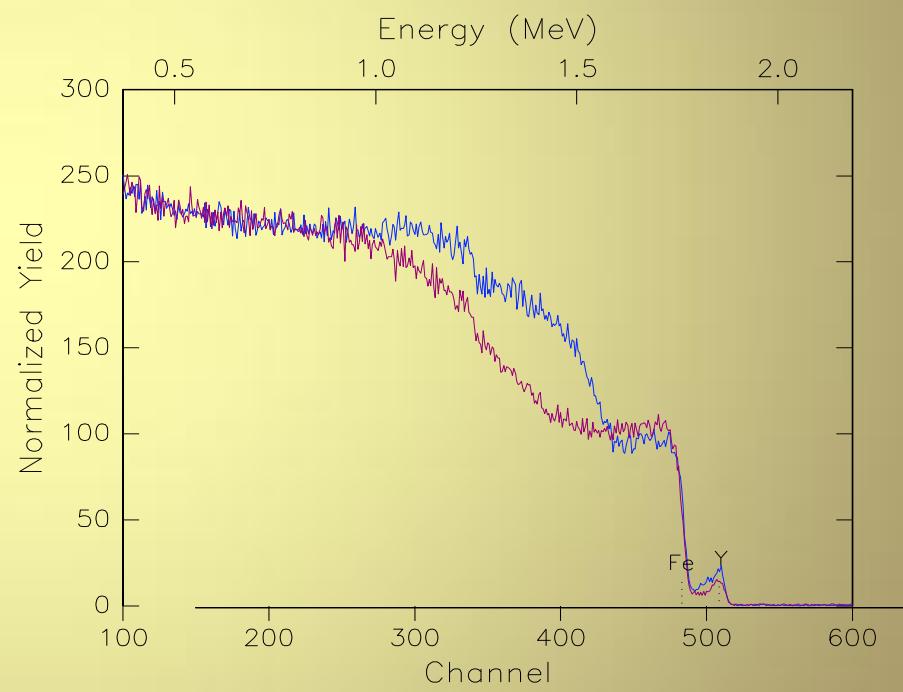
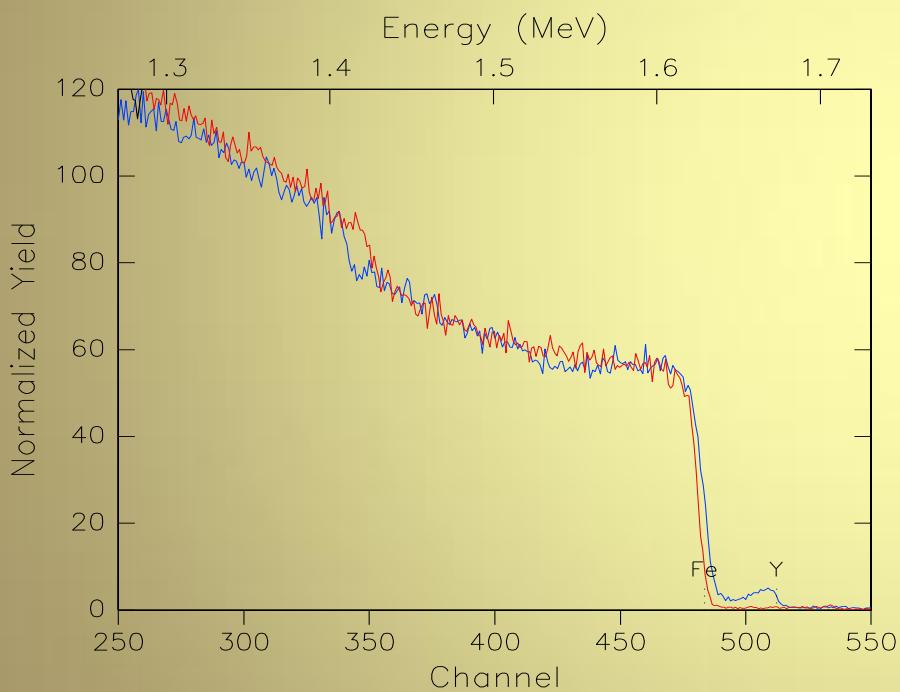


The H.V.E. 5.5 MV Tandem vdG accelerator of the NCSR *Demokritos* (Athens) and the utilized Charles Evans & Assoc. scattering chamber

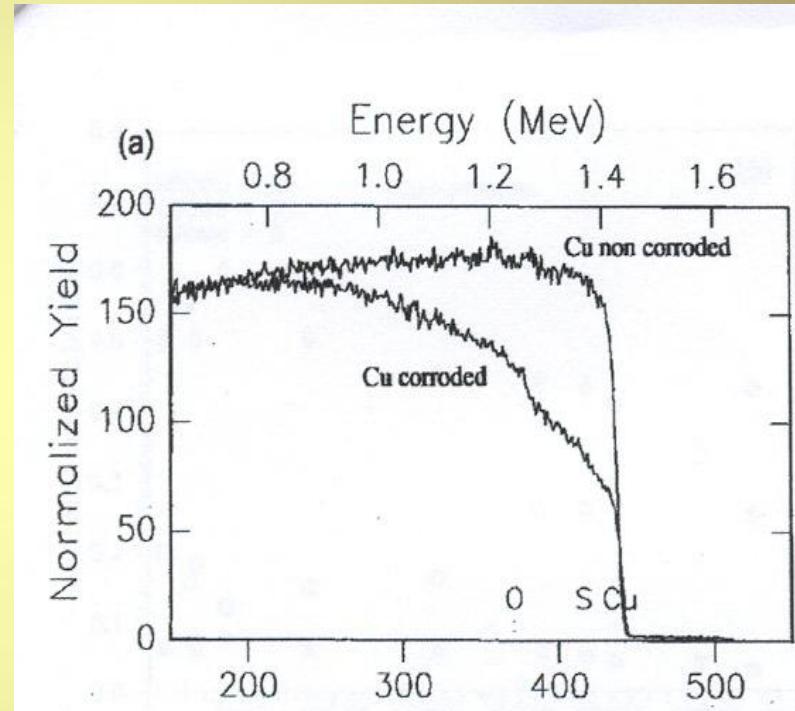
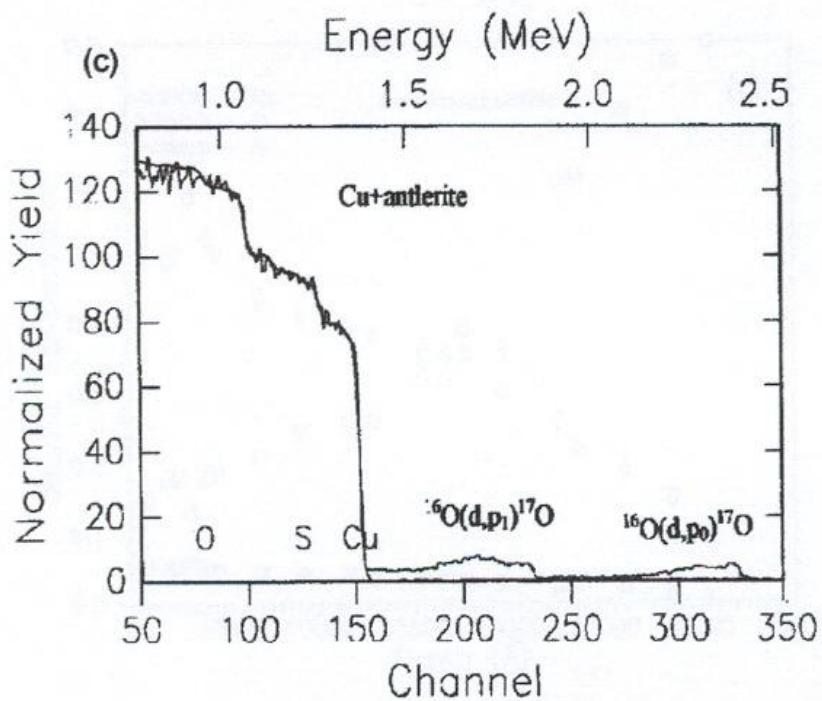


Microbeam  
Line

RBS-spectra; a) Y-implanted steel (40keV) and non implanted steel oxidised at 900 °C, b) Y-implanted steel (55 and 80 keV respectively) oxidised at 900 °C



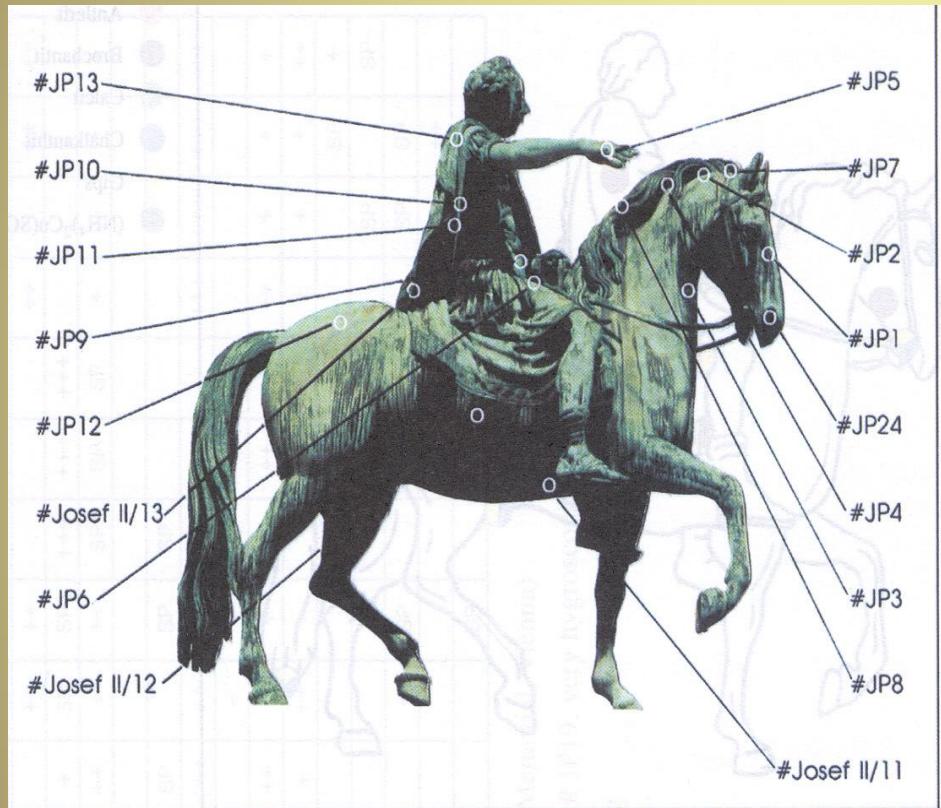
# Application of RBS to Corrosion Studies



From:

F. Noli, P. Misaelides, A. Hatzidimitriou, E. Pavlidou and M. Kokkoris, Investigation of artificially produced and natural copper patina layers, J. Mat. Chem. 13(2003)114

# Investigation of Cu-Patinas (natural and synthetic)

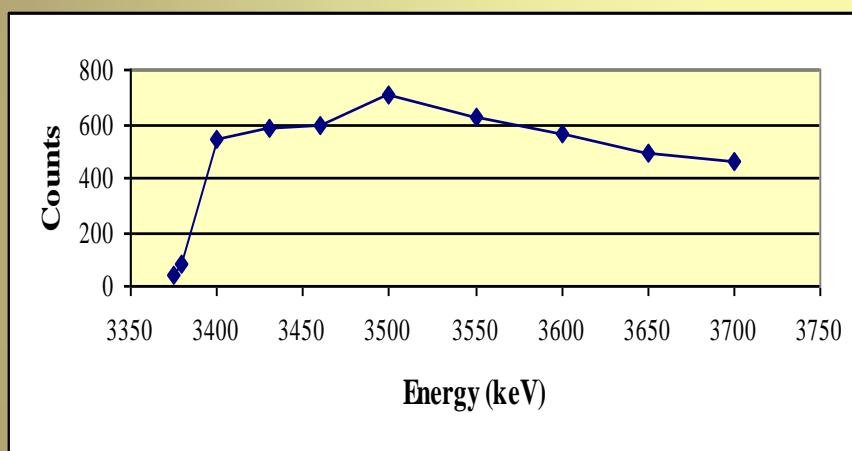


Work performed within the frame of  
the ENV4-CT95-0098 Project

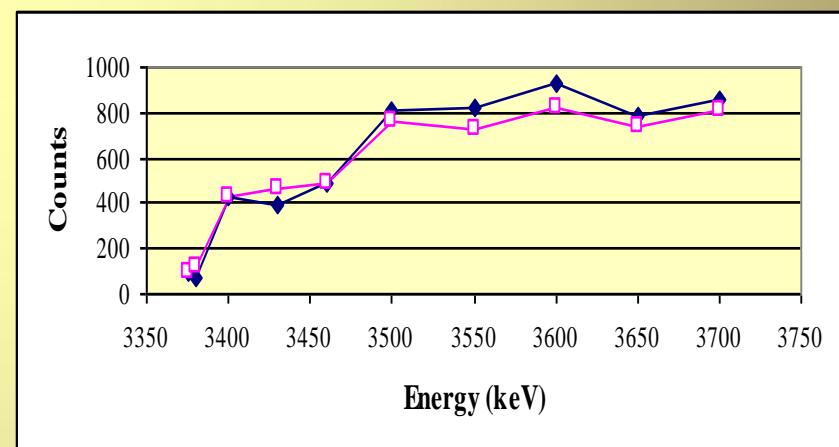
# Sulfur distribution in patina layer determined by means of $^{32}\text{S}(\text{p},\text{p}'\gamma)^{32}\text{S}$ nuclear reaction

( $E_{\text{res}} = 3716 \text{ keV}$ ,  $E_{\gamma} = 2230 \text{ keV}$ ,  $d\sigma(E, 90^\circ)\Gamma/d\Omega = 48.10 \text{ mb/sr}$  from C. Tsartsarakos, P. Misaelides and A. Katsanos, Nucl. Instr. and Meth. B45(1990)33)

## Natural patina (Vienna Hofburg)



Corroded (♦) and non-corroded (□)  
mixed patina consisting of antlerite  
 $(\text{CuSO}_4 \cdot 2\text{Cu(OH)}_2)$  + brochantite  
 $(\text{CuSO}_4 \cdot 3\text{Cu(OH)}_2)$  + chalcanthite  
 $(\text{CuSO}_4 \cdot 5\text{H}_2\text{O})$



From

F. Noli, P. Misaelides, M. Kokkoris, Investigation of natural and artificially produced copper patina layers using ion-beam analysis techniques, Proc. of the Nuclear and Related Techniques Conference, La Habana, Cuba, October 2003

# • Radiochemical techniques

## Our aim:

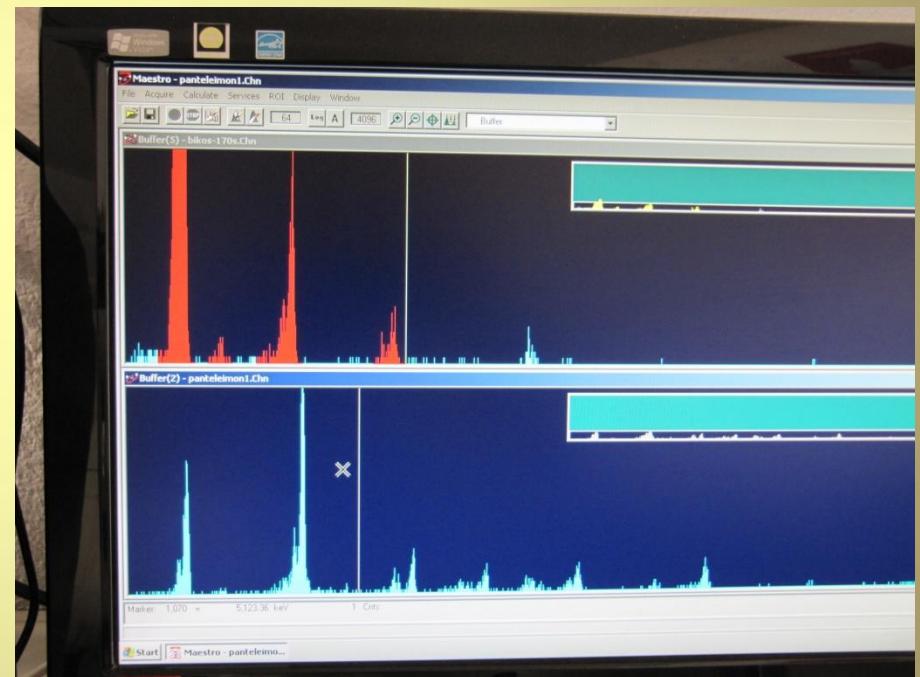
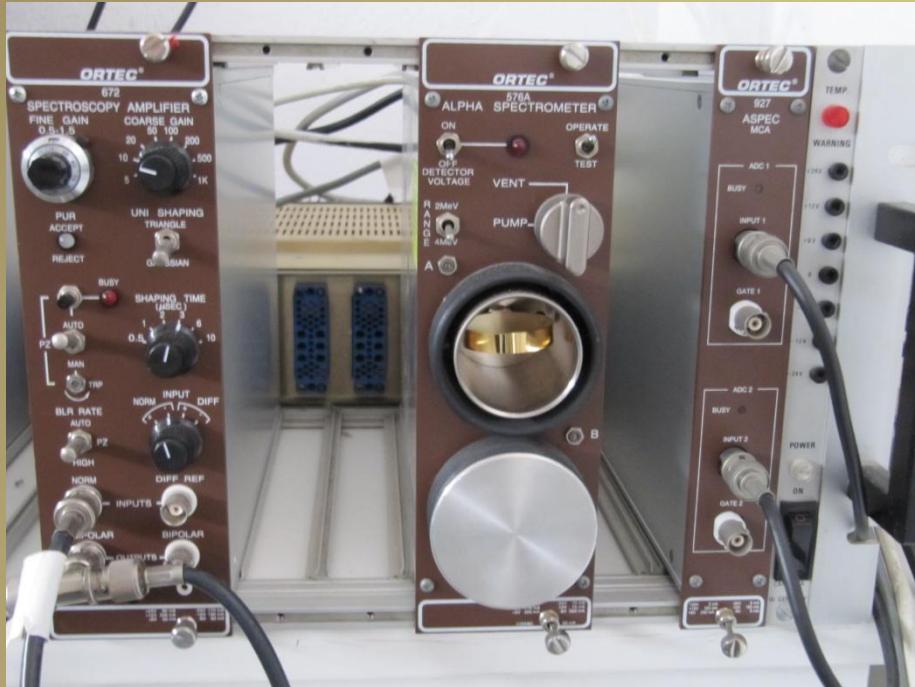
- measurement of the natural radioactivity using  $\gamma$ -ray spectroscopy
- determination of alpha-emmiters (radionuclides: U-238, U-235, U-234, Ra-226, Ra-224) using  $\alpha$ -ray spectroscopy

## The materials

-Environmental samples (waters, soils, sediments, air filters etc.)







*Thank you*